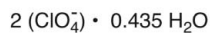
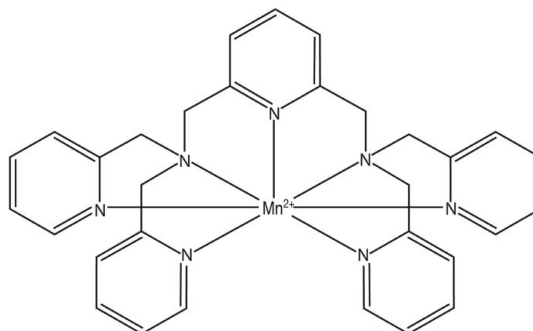


**{2,6-Bis[bis(2-pyridylmethyl)aminomethyl]pyridine}-  
manganese(II) bis(perchlorate) 0.435-hydrate**Alan Hazell<sup>a</sup> and Hans  
Toftlund<sup>b\*</sup><sup>a</sup>Department of Chemistry, Aarhus University,  
Langelandsgade 140, 8000 Århus C, Denmark,  
and <sup>b</sup>Department of Physics and Chemistry,  
University of Southern Denmark, Campusvej 55,  
5230 Odense M, Denmark

Correspondence e-mail: ach@chem.au.dk

**Key indicators**Single-crystal X-ray study  
 $T = 100\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$   
Disorder in solvent or counterion  
 $R$  factor = 0.035  
 $wR$  factor = 0.037  
Data-to-parameter ratio = 17.5For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.The title compound,  $[\text{Mn}(\text{C}_{31}\text{H}_{31}\text{N}_7)](\text{ClO}_4)_2 \cdot 0.435\text{H}_2\text{O}$ , contains a high-spin  $\text{Mn}^{\text{II}}$  atom which is coordinated by seven N atoms at the corners of a distorted pentagonal bipyramid.Received 19 December 2006  
Accepted 22 December 2006**Comment**The title compound,  $[\text{Mn}(\text{BPATPA})](\text{ClO}_4)_2 \cdot 0.435\text{H}_2\text{O}$ , (I), is isostructural with  $[\text{Co}(\text{BPATPA})](\text{ClO}_4)_2 \cdot 0.365\text{H}_2\text{O}$ , described in our preceding report (Hazell & Toftlund, 2007). In (I), the  $\text{Mn}^{\text{II}}$  atom is coordinated by seven N atoms which form a distorted pentagonal bipyramid (Fig. 1 and Table 1), with atoms N21 and N31 in the axial positions. The axial  $\text{Mn}-\text{N}_{sp^2}$  bonds are longer than the equatorial  $\text{Mn}-\text{N}_{sp^2}$  bonds and are similar in length to the  $\text{Mn}-\text{N}_{sp^3}$  bonds (atoms N1 and N9). The mean  $\text{Mn}-\text{N}$  bond length of  $2.351\text{ \AA}$  compares with the mean  $\text{Fe}-\text{N}$  bond length of  $2.313\text{ \AA}$  in  $[\text{Fe}(\text{BPATPA})](\text{ClO}_4)_2 \cdot \text{H}_2\text{O}$  (Lonnon *et al.*, 2002) and the mean  $\text{Co}-\text{N}$  bond length of  $2.274\text{ \AA}$  in  $[\text{Co}(\text{BPATPA})](\text{ClO}_4)_2 \cdot 0.365\text{H}_2\text{O}$  (Hazell & Toftlund, 2007). Thus, the mean  $M-\text{N}$  bond length decreases regularly with increasing number of unpaired  $d$  electrons on  $M$  in these high-spin complexes.

(I)

As is the case for the  $\text{Co}^{\text{II}}$  complex, the water molecule is included with fractional site occupancy and is hydrogen bonded to two O atoms of two perchlorate anions (Table 2).**Experimental**The BPATPA ligand (0.25 g, 0.5 mmol) was dissolved in methanol (10 ml) and the solution was flushed with dinitrogen for 5 min.  $\text{Mn}(\text{ClO}_4)_2 \cdot 6\text{H}_2\text{O}$  (0.18 g, 0.5 mmol) was added and the solution was stirred with continuous bubbling of dinitrogen for another 15 min. The solution was then left to stand in a closed vessel and crystals of (I) formed overnight.

## Crystal data

[Mn(C<sub>31</sub>H<sub>31</sub>N<sub>7</sub>)](ClO<sub>4</sub>)<sub>2</sub>·0.435H<sub>2</sub>O $M_r = 763.34$ Monoclinic,  $P2_1/c$  $a = 14.9038$  (9) Å $b = 12.3825$  (8) Å $c = 18.647$  (1) Å $\beta = 108.733$  (1)° $V = 3258.9$  (4) Å<sup>3</sup> $Z = 4$  $D_x = 1.556$  Mg m<sup>-3</sup>Mo  $K\alpha$  radiation $\mu = 0.63$  mm<sup>-1</sup> $T = 100$  K

Block, colourless

 $0.42 \times 0.34 \times 0.32$  mm

## Data collection

Siemens SMART CCD area-detector diffractometer

 $\omega$  scans

Absorption correction: integration (XPREP; Siemens, 1995)

 $T_{\min} = 0.757$ ,  $T_{\max} = 0.844$ 

35942 measured reflections

9357 independent reflections

7918 reflections with  $I > 3\sigma(I)$  $R_{\text{int}} = 0.038$  $\theta_{\text{max}} = 29.8^\circ$ 

## Refinement

Refinement on  $F$  $R[F^2 > 3\sigma(F^2)] = 0.035$  $wR(F^2) = 0.037$  $S = 1.04$ 

7918 reflections

452 parameters

H-atom parameters constrained

$$w = 1/[\sigma_{\text{cs}}(F^2 + B) + (1 + A)F^2]^{1/2} - |F|^2,$$

where  $A = 0.03$  and  $B = 2.0$  $(\Delta/\sigma)_{\text{max}} < 0.001$  $\Delta\rho_{\text{max}} = 0.60$  e Å<sup>-3</sup> $\Delta\rho_{\text{min}} = -0.39$  e Å<sup>-3</sup>

Table 1

Selected bond lengths (Å).

Mn1—N1	2.397 (1)	Mn1—N21	2.431 (1)
Mn1—N9	2.409 (1)	Mn1—N31	2.350 (1)
Mn1—N10	2.290 (1)	Mn1—N41	2.279 (1)
Mn1—N11	2.306 (1)		

Table 2

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O9—H9A $\cdots$ O5	0.85	1.92	2.768 (4)	180
O9—H9B $\cdots$ O4	0.85	2.31	3.165 (4)	180

H atoms bound to C atoms were placed in calculated positions, with  $C-H = 0.95$  Å, and refined using a riding model, with  $U_{\text{iso}}(\text{H}) =$

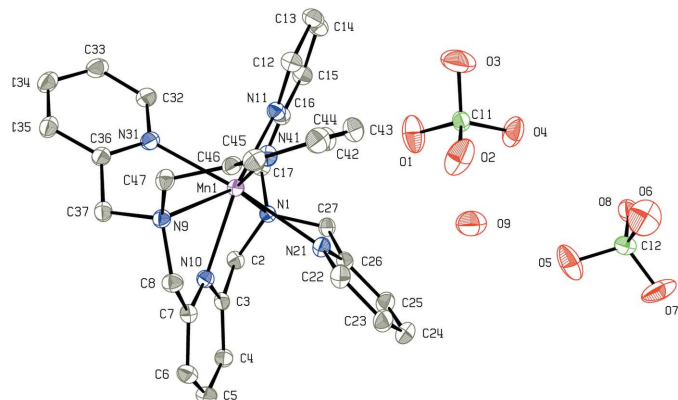


Figure 1

The molecular structure of (I), showing displacement ellipsoids at the 50% probability level. H atoms have been omitted.

$1.2U_{\text{eq}}(\text{C})$ . The H atoms of the water molecule were placed along the vector from O9 to the nearest hydrogen-bond acceptor O atom, with  $O9-H = 0.85$  Å. They were allowed to ride during subsequent refinement, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{O})$ . The refined site occupancy factor of atom O9 is 0.435 (5).

Data collection: SMART (Siemens, 1995); cell refinement: SAINT (Siemens, 1995); data reduction: SAINT; program(s) used to solve structure: SIR97 (Altomare *et al.*, 1997); program(s) used to refine structure: KRYSTAL (Hazell, 1995); molecular graphics: ORTEPIII (Burnett & Johnson, 1996); software used to prepare material for publication: KRYSTAL.

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